# Consumer Confidence Report Certification Form (to be submitted with a copy of the CCR)

Water System Name:		m Name: <u>Ke</u>	nwood Village Water Company							
Water	r Systei		4910025							
Furth	c/y/ er, the	, 20/6 (date) to custing (date) to custing (date)	by certifies that its Consumer Confidence Report was distributed on stomers (and appropriate notices of availability have been given). information contained in the report is correct and consistent with the sly submitted to the Department of Public Health.							
Certif	fied by:	Name:	Julie Cavaz							
		Signature:	garhelemm							
		Title:	Assistant Manager							
		Phone Number:	(701):539-6397 Date: 7/1/16							
	CCR metho	was distributed by maids used:  July 1, 201  I faith" efforts were use wing methods:	Lor other direct delivery methods. Specify other direct delivery							
		•	he CCR on the Internet at www.							
		Mailing the CCR to pos	postal patrons within the service area (attach zip codes used)							
		Advertising the availab	ility of the CCR in news media (attach copy of press release)							
		Publication of the CC published notice, include	R in a local newspaper of general circulation (attach a copy of the ling name of newspaper and date published)							
•		Posted the CCR in pub	ic places (attach a list of locations)							
	×	Delivery of multiple coas apartments, business	opies of CCR to single-billed addresses serving several persons, such es, and schools							
٠		Delivery to community	unity organizations (attach a list of organizations)							
	For systems serving at least 100,000 persons: Posted CCR on a publicly-accessible internet site a the following address: www.									
	For privately-owned utilities: Delivered the CCR to the California Public Utilities Commission									

## **2015 Consumer Confidence Report**

## Kenwood Village Water Co.

#### July, 2016

We test the drinking water quality for many constituents as required by State and Federal Regulations. This report shows the results of our most recent monitoring.

Name & location of source(s): Well #K-1 Greene Street (95% of production)

Well #K-2 between Greene Street & Los Guillicos (5% of production)

Type of water source(s) in use: Both Well #K-1 and Well #K-2 are ground water sources.

<u>Vulnerability Summary</u>: Well #K-1 and #K-2 are considered most vulnerable to the following activities not associated with any detected contaminants: Gas station, historic dumps/landfills, high septic system density. More information is available at the State Department of Public Health, telephone (707) 576-2145.

For more information, contact: Karen Ball Phone: (707) 539 - 6397

#### Terms used in this report:

Maximum Contaminant Level (MCL): The highest level of a contaminant allowed in drinking water. Primary MCLs are set as close to the PHGs (or MCLGs) as is economically and technologically feasible. Secondary MCLs are set to protect the odor, taste and appearance of drinking water.

Public Health Goal (PHG): The level of a contaminant In drinking water below which there is no known or Expected risk to health. The California Environmental Protection Agency sets PHG's.

Primary Drinking Water Standards (PDWS): MCL for contaminants that affect health along with monitoring, reporting and water treatment requirements. Maximum Contaminant Level Goal (MCLG): The Level of a contaminant in drinking water below which There is no known or expected risk to health. MCLGs Are set by the United States Environmental Protection Agency (USEPA).

Secondary Drinking Water Standards (SDWS): MCLs for contaminants that affect taste, odor or appearance of drinking water. Contaminants with SDWSs do not affect the health at the MCL levels.

Regulatory Action Level (AL): The Concentration of which, if exceeded, triggers treatment or other requirements which a water system must follow.

ND: not detectable at testing limit

**Ppm:** parts per million or milligrams per liter (mg/L)

ppb: parts per billion or micrograms per liter (ug/L)

ppt: parts per trillion or nanograms per liter (ng/L)

pCi/L: picocuries per liter (a measure of radiation)

The sources of drinking water (both tap and bottled waters) include rivers, lakes, streams, ponds, reservoirs, springs and wells. As water travels over the surface of the land or through the ground, it dissolves naturally occurring minerals and, in some cases, radioactive material. It can, also, pick up substances resulting from the presence of animals or from human activity.

#### Contaminants that may be present in source water include:

Microbial contaminants, such as viruses and bacteria, which may come from sewage treatment plants, septic systems, agricultural livestock operations and wildlife.

Inorganic contaminants, such as salts and metals, that can be naturally-occurring or results from urban storm water runoff, industrial or domestic wastewater discharges, oil and gas production, mining or farming.

Pesticides and herbicides, which may come from a variety of sources such as agriculture, urban storm water runoff and residential uses.

Organic chemical contaminants, including synthetic and volatile organic chemicals, that are byproducts of industrial processes and petroleum production, and can also come from gas stations, urban storm water runoff and septic systems.

Radioactive contaminants, which can be naturally occurring or be the result of oil and gas production and mining activities.

In order to ensure tap water is safe to drink, USEPA and the State of California Department of Health Services prescribe regulations that limit the amount of contaminants in water provided by public water systems. Department regulations also establish limits for contaminants in bottled water that must provide the same protection for public health.

Tables list all the drinking water contaminants detected during the most recent sampling for the constituent. The presence of these contaminants in the water does not necessarily indicate that the water poses a health risk. The Department requires us to monitor for certain contaminants less than once per year because the concentrations of these contaminants are not expected to vary significantly from year to year. Some of the data, though representative of water quality, is more than one year old.

SA	MPLI	NG RI	ESULS S	HOWING	THE	DETEC	TION OF	COLIFO	RM BACTERIA
MICROBIOLOGICAL CONTAMINANTS ( to be completed only was a detection of ba	No		No. of Ionths in Violation	MCL		L	MCLG	Typical Source of Bacteria	
Total Coliform Bacteria (24/yr '15	ı '	mo.)		More than 1 sample in A month with a detection			0	Naturally present in the environment	
Fecal Coliform or E. coli (24/yr '15)		уе	the ar)	0	A routine sample and A repeat sample Detect total coliform and either sample And either sample also detects Fecal Coliform or E. coli			0 OF LEAD	Human and animal fecal waste
Lead and Copper	No. Sar		90 % percentile level detected	No. Sites exceeding AL	AL	PHG MCLG	Typical Source of Contaminant		
Lead (ppm) 8/13			dustrial manufacturers;						
Copper (ppm) 8/13	10	0.0720	0	1.3	0.17			ehold water plumbing systems; erosion of from wood preservatives	

## SAMPLING RESULTS FOR SODIUM AND HARDNESS

#### WELL K-1 WELL K-2

Chemical or Constituent (and reporting units)	Sample Data	Level Detected	Level Detected	MCL	PHG (MCLG)	Typical Source of Contaminant
Sodium (mg/L)	6/15	13	43	None	None	Generally found in ground and surface water.
Total Hardness	6/15	180	110	None	None	Generally found in ground and surface water.
Fluoride	6/15	0.18	0.33	2	None	Eroston of natural deposits, discharge from lettilizer and aluminum factories, a water additive which promotes strong teeth
Calcium	6/15		14	None	None	Erosion of natural deposits
Magnesium	6/15	36	17	None	None	Leaching from natural deposits

## DETECTION OF CONTAMINANTS WITH A PRIMARY DRINKING WATER STANDARD

		WELL K-1	WELL K-2			The second secon
Chemical or Constituent ( and reporting units)	Sample Date Every 3 yrs.	Level Detected	Level Detected	MCL	PHG	Typical Source of Contaminant
Aluminum (ppm)	6/15	<0.50	0.110	1	0.6	Erosion of natural deposits, residue from some surface water treatment processes.
Antimony (ppb)	6/15	<6	<6	6	N/A	Discharge from petroleum refineries, fire retardants, ceramics, electronics, solder.
Arsenic (ppb)	6/15	<2	<2	50	N/A	Erosion of natural deposits, runoff from orchards, glass and electronics production wastes.
Barium (ppb)	6/15	<0.10	<0.10	1	N/A	Discharge of oil drilling wastes and from metal refineries; erosion of natural deposits.
Beryllium (ppm)	6/15	<1	<1	4	N/A	Discharge from metal refineries, coal burning factories, and electrical, aerospace, and defense industries.
Cadmium (ppb)	6/15	<1	<1	5	.07	Internal corrosion of galvanized pipes, erosion of natural deposits, discharge from electroplating and industrial chemical factories and metal refineries, runoff from waste batteries and paints.
Chromium (ppb)	6/15	9.6	<1.0	50	N/A	Discharge from steel and pulp mills and chrome plating, and erosion of natural deposits.
Hexavalent Chromium (ppb)	2015 Ave.	9.89	ND	10	0.02	Discharge from electroplating, leather tanneries, wood preservation, chemical synthesis, refractory production, textile manufacturing, and erosion of natural deposits.
Mercury (ppb)	6/15	<1	<1	2	0.12	Erosion of natural deposits; discharge from refineries and factories; runoff from landfills, runoff from cropland.
Nitrate (ppm) (As NO3)	6/15	14	>2	45	45	Runoff from fertilizer usage, leaching from septic tanks, sewage.
Nitrate (as N)	6/15	<400		1000	N/A	Runoff from fertilizer usage, leaching from septic tanks, sewage.
Nickel(ppb)	6/15	<10	<10	100	12	Erosion of natural deposits, discharge from metal factories.
Selemium (ppb)	6/15	<5	<5	50	NA	Discharge from petroleum, glass, and metal refineries; erosion of natural deposits; discharge from mines and chemical manufactures; runoff from livestock lots (feed additive)
Thallium (ppb)	6/15	<l< td=""><td><l< td=""><td>2</td><td>0.1</td><td>Leaching from ore processing sites; discharge from electronics, glass and drug factories.</td></l<></td></l<>	<l< td=""><td>2</td><td>0.1</td><td>Leaching from ore processing sites; discharge from electronics, glass and drug factories.</td></l<>	2	0.1	Leaching from ore processing sites; discharge from electronics, glass and drug factories.
Zine (ppm)	6/15	<0.50	< 0.50	5	NA	Runoff/leaching from natural deposits; industrial wastes.
2,4,5-T	6/15	<2.0	<2.0	10	N/A	Some people who drink water containing 2,4,5-T in excess over Many years can experience liver problems.

#### **DETECTION OF RADIOACTIVE CONTAMINANTS**

Well K-1 Well K-2 Chemical or Sampling Range Range MCL Typical Source of Constituent Date Detected Detected Contaminant Gross Alpha 4 qtrs/15 0.0 - .858 15 Decay of natural and (pCi/L) manmade deposits Gross Alpha 4 qtrs/07 .0.0 -1.06 15 Decay of natural and (pCi/L) manmade deposits

DETECTION CONTAMINANTS - SECONDARY DRINKING WATER STANDARD WELL K-1 WELL K-2									
Chemical or Constituent (and reporting units)	Sampling Date	Level Detected	Level Detected	MCL	Typical Source of Contaminant				
Sulfate (ppm)	6/15	16	3.2	500	Runoff/leaching from natural deposits' industrial				
Chloride (ppm)	6/15	10	10	500	Runoff/leaching from natural deposits; seawater influence				
Specific Conductance	6/15	440	420	1600	Substances that form ions when in water; seawater influence				
Foaming Agents (MBAS) (ppb)	6/15	< 0.05	< 0.05	500	Municipal and industrial waste discharges				
Copper (ppm)	6/15	<.050	<.050	1.0	Internal corrosion of household plumbing systems; erosion of natural deposits.				
Iron (ppb)	6/15	< 100	<100	300	Leaching from natural deposits; industrial wastes				
Manganese (ppb)*	6/15	< 20	*330	50	Leaching from natural deposits				
Silver (ppb)	7/15	< 10	< 10	100	Industrial discharges				
РН	6/15	7.3	7.6	None	Measure of alkalimity or acidity of water				
Apparent Color	6/15	5.0	35	15	ivaturally occurring organic materials				
Lab Turbidity	6/15	0.35	0.30	5	Soil runoff				

Any violation of an MCL or AL is asterisked. Additional information regarding the violation is provided on the last page. There are no PHGs or MCLGs for constituents with secondary drinking water standards these are not health-based levels, but set on the basis of esthetics.

Well K-2

Well K-1

<4

#### **Additional Contaminants**

Perchlorate (ppb)

		11 CH 12-1	VV CII XX-2		
Chemical or	Sample Date	Level	Level	MCL	Typical Source
constituent		Detected	Detected		of Contaminants
Methyltert butyl ether (MTBE)*	6/12	0	0	5	Discharge from petroleum refineries & Industrial chemical factories.
Trihalomethanes (total)	8/15	3.3	.5	100	By product of drinking water chlorination
Haloacetic acids (5)	8/15	1.43	ND	60	
Asbestos	8/12	ND	ND	7	Internal corrosion of asbestos cement water mains, erosion of natural deposits
Boron (ppb)	2x/03	<100 (Ave)	145 (Ave)	None	AL 1000 Erosion of natural deposits
Vanadium (ppb)	2x/03	5.35 (Ave)	4.70 (Ave)	None	AL 50 Erosion of natural deposits
1,2,3- trichloropropane	8/09	<0.50	<0.50	AL 0.50	Some people who use water in excess of action levels over many years have

## **Additional General Information On Drinking Water**

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6

an increased risk of cancer, based on

studies in laboratory animals.
Runoff/leaching of natural deposits

and industrial manufacturers.

All drinking water, including bottle water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate the water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the USEPA Safe Drinking Water Hotline (1-800 - 426 - 4791)

Some people may be more vulnerable to contaminants in drinking water than the general population. Imo-compromised persons such as those with cancer, undergoing chemotherapy; those who have undergone organ transplants; those with HIV/AIDS or other immune system disorders; some elderly, and infants can be particularly at risk from infections. These people should seek advice about drinking water from their health care providers. USEPA/Centers for Disease Control (CDC) guideline on appropriate means to lessen the risk of infection by Cryptosporidium and other microbial contaminants are available from the Safe Drinking Water Hotline(1-800-426-4791)

#### OVER 80 ADDITIONAL CHEMICALS WERE TESTED FOR AND NOT DETECTED

4/12

\* The standard for manganese that exceeds the MCL is a secondary one. Although manganese poses no health problem, it can discolor water. Manganese is a naturally occurring mineral that is an essential nutrient.